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Chapter 12 Multiple Choice Identify the letter of the choice that best completes the statement or answers the question. ____ 1. The first step in most stoichiometry problems is to ____ a. add the coefficients of the reagents c. convert given quantities to volumes b. convert given quantities to moles d. convert given quantities to masses ____ 2.

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Chapter 12 Stoichiometry Test B Answers

Chapter 12 Stoichiometry 127 SECTION 12.1 THE ARITHMETIC OF EQUATIONS (pages 353-358) This section explains how to calculate the amount of reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

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Part 12.11A : First, we will calculate the theoretical yield based on the stoichiometry. Step 1: List the known quantities and plan the problem. Known : given: mass of KClO 3 = 40.0 g; molar mass KClO 3 = 122.55 g/mol; molar mass O 2 = 32.00 g/mol; Unknown : theoretical yield O 2 = ? g

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