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Study Guide Key Concepts 9.1 Naming Ions • When the metals in Groups 1A, 2A, and 3A lose electrons, they form cations with posi- ... an interactive guided tutorial. Chemical Names and

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When the anion name ends in -ide, the stem of the anion name begins with the prefix hydro- and ends with -ic. The name is then followed by the word acid. 2. When the anion name ends in -ite, the anion ends with -ous, followed by the word acid. 3. When the anion name ends in -ate, the anion name ends with -ic, followed by the word acid.

CHAPTER 9: Chemical Names and Formulas

The name of a base is the name of the cation followed by the name of the anion (hydroxide). The formula of a base is written by showing the number of hydroxide ions needed to balance the positive charge on the cation. After reading Lesson 9.4, answer the following questions. Names and Formulas of Acids hydrogen1.

Chemical Names and Formulas - AP Biology

86 Guided Reading and Study Workbook 6. Complete the table using the law of multiple proportions. Practicing Skills: Naming Chemical Compounds (pages 276-277) 7. How can a flowchart help you to name chemical compounds? 8. Use the flowchart in Figure 9.20 on page 277 to write the names of the following compounds: a. CsCl _____

Name Date Class CHEMICAL NAMES AND FORMULAS 9

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Chemical Formula Writing Worksheet Two Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the “zinc” cation and the “chloride” anion, so you should write “ZnCl₂”, as shown. Cations

STUDY GUIDE: Naming & Formulas of Ionic Compounds

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The general chemical formulas of acids is H_nX. X is a monatomic or polyatomic anion n is a subscript indication the number of hydrogen ions combined with the anion. Naming Acids. Three rules are used to name acids. The name depends on the name of the anion and its suffix (-ide, -ite, -ate) 1. a. When the name of the anion ends in -ide

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Chapter 9 Chemical Names and Formulas Section 9.1 Naming Ions

It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction. Writing Chemical Equations (pages 321-323) 1. ... 114 Guided Reading and Study Workbook CHAPTER 11, Chemical Reactions(continued) 9.

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321-329)

24.95. 114 Guided Reading and Study Workbook CHAPTER 11, Chemical Reactions Name Date Class Chemical Reactions Guided Reading and Study. Chemical reactions occur when atoms are separated, joined, or rearranged, Chapter 4 Atomic Structure 33 Page 6/15

Chapter 11 Chemical Reactions Guided Reading Answers

Name Chemical Reactions Guided Reading and Study 19. Tell why this chemical equation is not balanced: $H_2 + O_2 \rightarrow H_2O$. Class 20. Write a balanced equation for this reaction: Oxygen reacts with hydrogen to yield water. Classifying Chemical Reactions (pp. 200-201) 21. In what three categories can chemical reactions be classified? 22.

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Name Date Class Chemical Reactions Guided Reading and Study Controlling Chemical Reactions (pp. 204-209) This section explains how energy is related to chemical reactions. It also describes how the rates of chemical reactions can be controlled. Use Target Reading Skills

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Name Date Class Cell Processes and Energy • Guided Reading and Study Chemical Compounds in Cells (continued) C) 4. Complete this concept map on organic compounds. Organic compounds contain include Carbohydrates 5. Compounds that do not contain carbon are called Carbohydrates O 6. A carbohydrate is made of carbon, hydrogen, and 7.

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Guided Reading Study Chemical Compounds in Cells

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C.P. Chemistry Test Chapter 6 Study Guide Covalent Bonding ... Name Atoms and Bonding Date Guided Reading and Study Class Covalent Bonds This section describes how chemical bonds form when two atoms share electrons. It also describes how electrons are shared unequally in some chemical bonds. Use Target

Covalent Bonding Guided And Study Workbook Answers

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